2013 WATER QUALITY REPORT

Consumer Confidence Report for the period of January 1 to December 31, 2013



CITY OF HORSESHOE BAY

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This report is intended to provide you with important information about your drinking water and the efforts made by the water system to provide safe drinking water.

Drinking water, including bottled water, may reasonably be expected to contain at least small amounts of some contaminants. The presence of contaminants does not necessarily indicate that water poses a health risk. More information about contaminants and potential health effects can be obtained by calling the EPA's Safe Drinking Water Hotline at (800) 426-4791.

For more information regarding this report contact:

Jeff Koska, Utilities Director Office: 830-598-9981

Email: city@horseshoe-bay-tx.gov

Este reporte incluye informacion importante sobre el agua para tomar. Para asistencia en espanol, favor de llamar al telefono (830) 598-8741.

The source of drinking water used by the City of Horseshoe Bay is Surface Water from Lake Lyndon B. Johnson (LBJ).

Information about Source Water Assessments

A Source Water Susceptibility Assessment for your drinking water was completed by the Texas Commission on Environmental Quality (TCEQ). This information describes the susceptibility and types of constituents that may come into contact with your drinking water source based on human activities and natural conditions. The information contained in the assessment allows us to focus source water protection strategies. Results indicate that some of your sources are susceptible to certain contaminants. The sampling requirements for your water system are based on this susceptibility and previous sample data. Any detections of these contaminants may be found in this Consumer Confidence Report. For more information on source water assessments and protection efforts at our system, contact Jeff Koska, Utilities Director.

- For more information about your sources of water, please refer to the Source Water Assessment Viewer available at the following URL:
 http://teog/appropers/php1/
 - http://tceq4apmgwebp1.tceq.texas.gov:8080/swav/Controller/index.jsp?wtrsrc=
- Further details about sources and source water assessments are available in Drinking Water Watch at the following URL: http://dww.tceq.state.tx.us/DWW/

SPECIAL NOTICE

The sources of drinking water (both tap water and bottled water) include rivers, lakes, streams, ponds, reservoirs, springs, and wells. As water travels over the surface of the land or through the ground, it dissolves naturally-occurring minerals and, in some cases, radioactive material, and can pickup substances resulting from the presence of animals or from human activity.

Contaminants that may be present in source water include:

- Microbial contaminants, such as viruses and bacteria, which may come from sewage treatment plants, septic systems, agricultural livestock operations, and wildlife;
- Inorganic contaminants, such as salts and metals, which can be naturally-occurring or result from urban storm water runoff, industrial or domestic wastewater discharges, oil and gas production, mining, or farming;

- Pesticides and herbicides, which may come from a variety of sources such as agriculture, urban storm water runoff, and residential uses;
- Organic chemical contaminants, including synthetic and volatile organic chemicals, which are by-products of industrial processes and petroleum production, and can also come from gas stations, urban storm water runoff, and septic systems; and,
- Radioactive contaminants, which can be naturally-occurring or be the result of oil and gas production and mining activities.

In order to ensure that tap water is safe to drink, the Environmental Protection Agency (EPA) prescribes regulations which limit the amount of certain contaminants in water provided by public water systems. Food and Drug Administration (FDA) regulations establish limits for contaminants in bottled water which must provide the same protection for public health.

Contaminants may be found in drinking water that may cause taste, color, or odor problems. These types of problems are not necessarily causes for health concerns. For more information on taste, odor, or color of drinking water, please contact the water department at 830-598-8741.

You may be more vulnerable than the general population to certain microbial contaminants, such as Cryptosporidium, in drinking water. Infants, some elderly or immunocompromised persons such as those undergoing chemotherapy for cancer; persons who have undergone organ transplants; those who are undergoing treatment with steroids; and people with HIV/AIDS or other immune system disorders, can be particularly at risk from infections. You should seek advice about drinking water from your physician or health care providers. Additional guidelines on appropriate means to lessen the risk of infection by Cryptosporidium are available from the Safe Drinking Water Hotline (800) 426-4791.

If present, elevated levels of lead can cause serious health problems, especially for pregnant women and young children. Lead in drinking water is primarily from materials and components associated with service lines and home plumbing. We are responsible for providing high quality drinking water, but we cannot control the variety of materials used in plumbing components. When your water has been sitting for several hours, you can minimize the potential for lead exposure by flushing your tap for 30 seconds to 2 minutes before using water for drinking or cooking. If you are concerned about the lead in your water, you may wish to have your water tested. Information on lead in drinking water, testing methods, and steps you can take to minimize exposure is available from the Safe Drinking Water Hotline or at http://www.epa.gov/safewater/lead.

Definitions (The following tables contain scientific terms and measures used in this report, some of which may require explanation.)

MCL

NA

Not Applicable

Maximum Contaminant Level

		feasible using the best available treatment technology.	
MCLG	Maximum Contaminant Level	The level of a contaminant in drinking water below which there is no known or expected risk to health. MCLGs	

The highest level of a contaminant that is allowed in drinking water. MCLs are set as close to the MCLGs as

Collection Date; the year in which sample(s) were collected.

	Goal	allow for a margin of safety.
MRDL	Maximum Residual Disinfectant Level	The highest level of a disinfectant allowed in drinking water. There is convincing evidence that addition of a disinfectant is necessary for control of microbial contaminants.

WINDL	Level	disinfectant is necessary for control of microbial contaminants.
MRDLG	Maximum Residual Disinfectant Level Goal	The level of a drinking water disinfectant below which there is no known or expected risk to health. MRDLGs do not reflect the benefits of the use of disinfectants to control microbial contaminants.

	Level Goal	not reflect the benefits of the use of disinfectants to control microbial contaminants.
AL	Action Level	The concentration of a contaminant which, if exceeded, triggers treatment or other requirements which a water system must follow.

ALG	Action Level Goal	The level of a contaminant in drinking water below which there is no known or expected risk to health. allow for a margin of safety.	ALGs

		allow for a margin of safety.
AVG	Average	Regulatory compliance with some MCLs are based on running annual average of monthly samples.

AVG	Average	Regulatory compliance with some MCLs are based on running annual average of monthly samples.
NTH	Nenhelometric Turhidity Units	A measure of turbidity

NTU	Nephelometric Turbidity Units	A measure of turbidity.		
mg/L	Milligrams per liter or parts per m	illion – or one ounce in 7.350	ug/L	Micrograms per liter or parts per billion – or one ounce in

ppm	gallons of water.	ppb	7,350,000 gallons of water.	
MAX	Maximum	MIN	Minimum	

YEAR

2013 REGULATED CONTAININANTS DETECT	
Coliform Bacteria	

MCLG

0

90th

Percentile

0.199

2.49

Highest Level

Detected

2

0.052

0.6

0.093

2.41

2013 UNREGULATED CONSTITUENTS DETECTED

Secondary Constituents and Other Unregulated Constituents

AVG

Level

0.1337

146

30.6

Date of Detection

02/19/2013

Lead and Cooper

Year

2012

2012

Year

2013

Year

2013

2013

Year

2013

2013

2013

2013

2013

Year

2013

2013

2013

Contaminant (Unit)

Maximum Residual Disinfectant Level

Free Chlorine / Chloramines (ppm)

Disinfectants and Disinfection By-Products Disinfectant s and Disinfection

Haloacetic Acids (HAA5) (ppb)

Total Trihalomethanes (TThm) (ppb)

Disinfectant (Unit)

By-Products (Unit)

Inorganic Contaminants

(Unit)

Contaminant

Arsenic (ppb)

Barium (ppm)

Fluoride (ppm)

(Measured as

Nitrogen) (ppm)

Selenium (ppb)

Constituent (Unit)

Aluminum (mg/L)

Bicarbonate (mg/L)

Calcium (mg/L)

Nitrate -

and taste of your water.

Cooper (ppm)

Lead (ppb)

MCL

1

Action Level

(AL)

1.3

15

AVG

Level

3.2

Highest Level

Detected

20

35

Range of Levels

Detected

1.59-1.74

0.0377-0.052

0.22-0.64

0.047-0.093

2.13-2.41

MIN

Level

0.0384

143

30.5

WATER QUALITY TEST RESULTS

Sites Over

ΑL

0

0

MIN

Level

2.0

Highest Number of Positive

1

MCLG

1.3

0

MAX

Level

4.8

MCL

10

2

10

50

Limit

0.2

NA

NA

Range of Levels

Detected

8.4-30.9

0-55.3

MCLG

0

2

10

50

MAX

Level

0.229

149

30.7

Many constituents (such as calcium, sodium, or iron) which are often found in drinking water can cause taste, color, and odor problems. The taste and odor constituents are called secondary constituents and are regulated by the State of Texas, not the EPA. These constituents are not causes for health concern. Therefore, secondaries are not required to be reported in this document but they may greatly affect the appearance

E.Coli or Fecal Coliform Bacteria - Not Detected

Likely Source of Contamination

plumbing systems.

Likely Source of Contamination

refineries; erosion of natural deposits

sewage; erosion of natural deposits

aluminum factories

erosion of natural deposits.

Likely Source of Chemical

Likely Source of Contamination

Erosion of natural deposits; runoff from orchards;

runoff from glass and electronics production wastes.

Discharge of drilling wastes; discharge from metal

Erosion of natural deposits; water additive which promotes strong teeth; discharge from fertilizer and

Runoff from fertilizer use; leaching from septic tanks,

Discharge from petroleum and metal refineries; erosion of natural deposits; discharge from mines.

Likely Source of Contamination

Erosion of natural deposits; leaching from

wood preservatives; corrosion of household

Corrosion of household plumbing systems,

Disinfectant used to control microbes

By-product of drinking water disinfection.

By-product of drinking water disinfection.

Naturally present in the environment

Violation

NO

MRDL

4.0

Violation

NO

NO

Violation

NO

NO

MRDLG

4.0

MCL

60

80

Violation

NO

NO

NO

NO

NO

Likely Source of Contamination

Abundant naturally occurring element.

Corrosion of carbonate rocks such as limestone.

naturally in limestone, gypsum, and fluorite.

The fifth most abundant metal in the Earth's crust; It occurs

013 REGULATED CONTAININANTS DETECTED	
oliform Bacteria	

Coliform Bacteria		

2013	Magnesium (mg/L)	24.1	23.9	24.3	NA	NA The eighth most abundant metal in the Earth's crust; it occurs naturally in deposits of magnesite, dolomite, and other minerals.				
2013	Manganese (mg/L)	0.011	0.001	0.021	.05	Occurs naturally in the environment as solids in soils and small particles in water.			t as solids in soils and small	
2013	Nickel (mg/L)	.0004	.0004 .0004		0.1	Eros	sion of natural o			
2013	2013 pH (units) 7.95		7.3	8.6	NA	Measure of corrosivity of water.				
2013	2013 Sodium (mg/L) 29		22.7	32.6	NA	Eros	sion of natural o			
2013	Sulfate (mg/L)	26.6	23.1	30.1	300	Naturally occurring; low to moderate concentrations of both chloride and sulfate ions add palatability to water.				
2013	Total Alkalinity (mg/L)	146	143	149	NA	Naturally occurring soluble mineral salts.				
2013	Total Dissolved Solids (mg/L)	292.5	271	314	1000	Total dissolved mineral constituents in water.				
2013	Total Hardness Calcium and Magnesium (mg/L)	175.5	175	176	NA	Two of the principal elements making up the earth's crust; its compounds, when dissolved, make the water hard. The presof calcium and magnesium in water is a factor contributing to		the water hard. The presence		
	mg/L conversion to grains/gal	10.25	10.22	10.28	NA	formation of scale and insoluble soap curds which are a means of clearly identifying hard water.				
2013	Zinc (mg/L)	.0021	.0021	.0021	5	Moderately abundant naturally occurring element; high resistance to atmospheric corrosion. Used in the metal industry.				
Turbidity is a measurement of the cloudiness of the water caused by suspended particles. We monitor it because it is a good indicator of water quality and the effectiveness of our filtration.										
Year	Turbidity Limit Treatment Techniqu			Level Detecte	•		Likely Source	of Contamination		
2013 Highest Single Measurement		1 NTU		0.8	8 NO Soil runoff		Soil runoff			
2013	Lowest Monthly % Meeting Limit		0.3 NTU		99.41%	NO Soil runoff		Soil runoff		
MESSAGES FROM THE CITY A health-related note about landscape irrigation. Your home's landscape irrigation system can be a potential contamination source for household potable water should a reverse flow occur. To protect your potable water from flow reversals, make sure you have installed an adequate backflow prevention device on the irrigation main line and have the device tested annually. We encourage public interest and participation in our community's decisions affecting drinking water. City Council Meetings are held monthly at City Hall. Meeting schedules are posted on the City's website or you may call City Offices to obtain the schedule.										
CITY OF HORSESHOE BAY – WATER CONSERVATION							013 WATE	R SYSTEM	USAGE DATA	
STAGE 3 – ONCE PER WEEK WATERING ALLOWED										
RESIDENTIAL CUSTOMERS Even Numbered Address Odd Numbered Address				ross	Water Produced in 2013:			642,081 million gallons (MG) or 2292.0 acre/ft		
		-	ending in 1,3,5,		Maximur	m M	onth:		August - 82.7 MG	
Water on Thursday from Water on Saturday from 7PM to 10AM on Friday 7PM to 10AM on Sunday				Minimum Month:			February - 27.9 MG			
·					Maximum Day :			August 11 th - 3.834 MG		
COMMERCIAL AND MULTI-FAMILY CUSTOMERS Water on Tuesday from 7PM to 10AM on Wednesday				Average Day:			1.80 MG			
To check current restriction status or for additional water conservation information please refer to the City's website at:			-	# of Service Connections:			4703			
			te at:	Water Going Toward Outside Use:				61%		
http://www.horseshoe-bay-tx.gov					Water Entering Sewer:			33%		

Constituent (Unit)

Aluminum (mg/L)

Bicarbonate (mg/L)

Calcium (mg/L)

Chloride (mg/L)

Year

2013

2013

2013

2013

AVG

Level

0.1337

146

30.6

42.0

MIN

Level

0.0384

143

30.5

41.9

MAX

Level

0.229

149

30.7

41.9

Limit

0.2

NA

NA

300

Likely Source of Contamination

Abundant naturally occurring element.

Corrosion of carbonate rocks such as limestone.

naturally in limestone, gypsum, and fluorite.

The fifth most abundant metal in the Earth's crust; It occurs

A chemical compound of chlorine used in water purification.